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P1) It is clear that? tells us the number of gas molecules inside the box. Get the total number of molecules by multiplying? by Avogadro's number. The moles will tell you the number of atoms/molecules in 1L of gas. The number of molecules tells you the number of atoms in a mole of gas. Summation of these two tells you the total number of atoms/molecules in 1L of gas. This number is the content of Avogadro's number. This question can be solved by just drawing a table. But in case of reference, this is a useful way to check accuracy of data entered in the table. Objective of Quantum Physics??. The central nervous system senses a temperature change of? degrees Celsius as? changes to? degrees Celsius. The hypothalamus responds by releasing corticotropin releasing hormone, which causes the pituitary gland to release adrenocorticotropic hormone. The adrenal glands start secreting steroid hormones, resulting in an increase in blood glucose levels to the?. Page No: 383 Solution :: P2) This question is not clear. Could you clarify it, please? P3) According to the textbook, all atoms are essentially neutral. So, the fact that the state of the atom is (electron# 1) should mean that it is in an excited state (?). P4) This is a consequence of (e2) and (e3) of the text. P5) Since the speed of light in a vacuum is independent of the emission frequency (and likewise for absorption) and the frequency of light in a medium and that in a vacuum are the same (due to Lorentz contraction), you can deduce that energy cannot be conserved in

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